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## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

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<b>(21) International Application Number:</b> PCT/GB00/01706 <b>(22) International Filing Date:</b> 3 May 2000 (03.05.00) <b>(30) Priority Data:</b> 9911437.3                      17 May 1999 (17.05.99)                      GB <b>(71) Applicant (for AE AU BB CA CY DM GB GD GH GM IE IL KE LC LK LS MN MW NZ SD SG SL SZ TT UG ZA ZW only):</b> UNILEVER PLC [GB/GB]; Unilever House, Blackfriars, London EC4P 4BQ (GB). <b>(71) Applicant (for all designated States except AE AU BB CA CY DM GB GD GH GM IE IL IN KE LC LK LS MN MW NZ SD SG SL SZ TT UG ZA ZW):</b> UNILEVER NV [NL/NL]; Weena 455, NL-3013 AL Rotterdam (NL). <b>(71) Applicant (for IN only):</b> HINDUSTAN LEVER LIMITED [IN/IN]; Hindustan Lever House, 165/166 Backbay Reclamation, Mumbai 400 020, Maharashtra (IN).		<b>(72) Inventors:</b> GRAINGER, David, Stephen; Unilever Research Port Sunlight, Quarry Road East, Bebington, Wirral, Merseyside CH63 3JW (GB). GREEN, Andrew; Unilever Research Port Sunlight, Quarry Road East, Bebington, Wirral, Merseyside CH63 3JW (GB). MOHAMMADI, Mansur, Sultan; Unilever Research Port Sunlight, Quarry Road East, Bebington, Wirral, Merseyside CH63 3JW (GB). ROTH, Stephane; Unilever Research Port Sunlight, Quarry Road East, Bebington, Wirral, Merseyside CH63 3JW (GB). THOMPSON, Laurence, Griffith; Unilever Research Port Sunlight, Quarry Road East, Bebington, Wirral, Merseyside CH63 3JW (GB). <b>(74) Agent:</b> ELLIOT, Peter, William; Unilever Plc, Patent Department, Colworth House, Sharnbrook, Bedford MK44 1LQ (GB). <b>(81) Designated States:</b> AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).  <b>Published</b> <i>With international search report.</i>
<b>(54) Title:</b> FABRIC SOFTENING COMPOSITIONS <b>(57) Abstract</b> <p>The invention provides a fabric softening composition comprising: i) at least one nonionic fabric softening agent, and ii) at least one anionic surfactant, and iii) at least one cationic polymer wherein the particles formed from i), ii) and iii) have a net negative charge and the composition comprises no more than 1 % by weight of non-polymeric cationic surfactant and/or cationic fabric softening compounds.</p>		

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FABRIC SOFTENING COMPOSITIONSTechnical Field

5

The present invention relates to fabric softening compositions, particularly to compositions that soften without adversely affecting the absorbency of the fabric and which deposit well onto the fabric without being  
10 detrimentally affected by anionic carry-over from the wash.

Background and Prior Art

Rinse added fabric softener compositions are well known.  
15 However, a disadvantage associated with conventional rinse conditioners is that although they increase the soft feel of a fabric they simultaneously decrease the fabric's absorbency. A decrease in the absorbency properties of a fabric means that its ability to take up water decreases.  
20 This is particularly disadvantageous with towels where the consumer requires the towel to be soft, and yet, have a high absorbency.

WO 98/16538 (Unilever) discloses fabric conditioning  
25 compositions comprising liquid or soft solid derivatives of a cyclic polyol or a reduced saccharide which give good softening but retain absorbency of the fabric.

EP 0 380 406 (Colgate-Palmolive) discloses detergent  
30 compositions comprising a saccharide or reduced saccharide ester containing at least one fatty acid chain.

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WO 95/00614 (Kao Corporation) discloses softening compositions comprising polyhydric alcohol esters and cationised cellulose.

- 5 DE 19732073 (Henkel) discloses nitrogen free rinse conditioners containing water, anionic surfactants and fatty materials.

- 10 US 5 447 643 (Hüls) discloses aqueous fabric softeners comprising nonionic surfactants and mono, di or tri fatty acid esters of certain polyols.

EP 607529 (Huels) discloses nonionic fabric softening agents stabilised by cationic colloids.

15

WO 96/15213 (Henkel) discloses textile softening agents containing alkyl, alkenyl and/or acyl group containing sugar derivatives, which are solid after esterification, in combination with nonionic and cationic emulsifiers.

20

A further problem associated with fabric softening agents that are not cationic in nature is that deposition onto a fabric is often inadequate which generally leads to softening results that are not as good as the consumer

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The principal advantages of the present invention include that excellent softening of the fabric is achieved without detriment to the absorbency of the fabric, the softening agent deposits well onto fabric and is not unduly adversely affected by anionic carryover from the wash. Furthermore the compositions are easily manufactured.

#### Definition of the Invention

10 Thus according to one aspect of the invention there is provided a fabric softening composition comprising:

- (i) at least one nonionic fabric softening agent and
- 15 (ii) at least one anionic surfactant, and
- (iii) at least one cationic polymer

wherein the particles formed from i), ii) and iii) have an overall net negative charge and the composition comprises no more than 1% by weight non-polymeric cationic surfactant and/or cationic fabric softening compounds.

It has been found, surprisingly, that these compositions provide an unexpected combination of simultaneous fabric softening and retention of absorbency and also deposit well onto the fabric without being detrimentally affected by anionic carry-over from the wash.

30 The invention also provides a method of depositing a nonionic fabric softening agent onto fabric from a fabric softening composition, comprising emulsifying the softening

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agent with an anionic surfactant and a cationic polymer to form a particle having an overall negative charge in the composition and treating said fabric with said composition.

- 5 The invention further provides a method of depositing a nonionic fabric softening agent onto fabric from a fabric softening composition comprising emulsifying the softening agent with an anionic surfactant and then post-dosing an aqueous solution of a cationic polymer to form a particle  
10 having an overall negative charge in the composition and treating said fabric with said composition.

In the compositions of the invention the particles formed from the fabric softening agent, the anionic surfactant and  
15 cationic polymer have an overall net negative charge. This is measured by Zeta potential measurements (e.g. as measured on a Malvern Instrument Zeta-Sizer).

It is particularly surprising that the particles deposit  
20 onto the fabric because of their overall net charge. Without wishing to be bound by theory it is believed that the above overall negatively charged particles have sufficient local positive charge associated with the polymer to allow them to deposit onto the surface of the fabric.

25

#### Detailed Description of the Invention

##### Fabric Softening Agents



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The compositions of the invention comprise at least one fabric softening agent chosen from nonionic fabric softeners.

- 5 The nonionic fabric softener may be any such suitable softener, but, particularly preferred nonionic softeners are the CPE and RSE compounds as defined herein.

10 In the context of the present invention the initials CPE or RSE stand for a liquid or soft solid derivative of a cyclic polyol or a reduced saccharide respectively which results from 35 to 100% of the hydroxyl groups of the cyclic polyol or reduced saccharide being esterified and/or etherified, the CPE or RSE having two or more ester or ether groups  
15 independently attached to a C<sub>8</sub> to C<sub>22</sub> alkyl or alkenyl chain.

The CPE or RSE used according to the invention does not have any substantial crystalline character at 20°C. Instead it is preferably in a liquid or soft solid state as herein  
20 defined at 20°C.

The liquid or soft solid (as hereinafter defined) CPEs or RSEs of the present invention result from 35 to 100% of the hydroxyl groups of the starting cyclic polyol or reduced  
25 saccharide being esterified or etherified with groups such that they are in the requisite liquid or soft solid state.

Typically the CPE's or RSE's have 3 or more ester or ether groups or mixtures thereof, for example 3 to 8, eg 3 to 5.  
30 Preferably the CPE or RSE has 4 or more ester or ether

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groups. It is preferred if two or more of the ester or ether groups of the CPE or RSE are independently of one another attached to a C<sub>8</sub> to C<sub>22</sub> alkyl or alkenyl chain. The C<sub>8</sub> to C<sub>22</sub> alkyl or alkenyl groups may be branched or linear  
5 carbon chains.

Preferably 35 to 85% of the hydroxyl groups of the cyclic polyol or reduced saccharide, most preferably 40 to 80%, even more preferably 45 to 75%, such as 45 to 70% are  
10 esterified or etherified.

Preferably the CPE or RSE contains 35% tri or higher esters, eg at least 40%.

15 CPEs are preferred for use with the present invention. Inositol is a preferred example of a cyclic polyol. Inositol derivatives are especially preferred.

In the context of the present invention the term cyclic  
20 polyol encompasses all forms of saccharides. Indeed saccharides are especially preferred for use with this invention. Examples of preferred saccharides from which the CPE's or RSE's may be derived are monosaccharides and disaccharides.

25 Examples of monosaccharides include xylose, arabinose, galactose, fructose, sorbose and glucose. Glucose is especially preferred. Examples of disaccharides include maltose, lactose, cellobiose and sucrose. Sucrose is  
30 especially preferred.

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An example of a reduced saccharide is sorbitan.

The liquid or soft solid CPE's or RSE's of the present invention can be prepared by a variety of methods well known to those skilled in the art. These methods include acylation of the cyclic polyol or reduced saccharide with an acid chloride; trans-esterification of the cyclic polyol or reduced saccharide fatty acid esters using a variety of catalysts; acylation of the cyclic polyol or reduced saccharide with an acid anhydride and acylation of the cyclic polyol or reduced saccharide with a fatty acid. Typical preparations of these materials are disclosed in US 4 386 213 and 14416/88 (Procter and Gamble).

If the CPE is a disaccharide it is preferred if the disaccharide has 3 or more ester or ether groups. Particularly preferred CPE's are esters with a degree of esterification of 3 to 5, for example, sucrose tri, tetra and penta esters.

Where the cyclic polyol is a reducing sugar it is advantageous if each ring of the CPE has one ether group, preferably at the C<sub>1</sub> position. Suitable examples of such compounds include methyl glucose derivatives.

Examples of suitable CPEs include esters of alkyl(poly)glucosides, in particular alkyl glucoside esters having a degree of polymerisation from 1 to 2.

The liquid or soft solid CPE's or RSE's of the present invention are characterised as materials having a

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solid:liquid ratio of between 50:50 and 0:100 at 20°C as determined by T<sub>2</sub> relaxation time NMR, preferably between 43:57 and 0:100, most preferably between 40:60 and 0:100, such as, 20:80 and 0:100. The T<sub>2</sub> NMR relaxation time is commonly used for characterising solid:liquid ratios in soft solid products such as fats and margarines. For the purpose of the present invention, any component of the NMR signal with a T<sub>2</sub> of less than 100 microsecond is considered to be a solid component and any component with T<sub>2</sub> is greater than 100 microseconds is considered to be a liquid component. For the CPE's and RSE's the tetra, penta etc prefixes only indicate the average degrees of esterification. The compounds exist as a mixture of materials ranging from the monoester to the fully esterified ester. It is the average degree of esterification which is used herein to define the CPE's and RSE's.

The HLB of the CPE or RSE is typically between 1 and 3.

Factors governing the suitability of the CPE's and RSE's are the presence and degree of branched chains, mixed chain lengths and the level of unsaturation.

It has been found that CPE's and RSE's having unsaturated or mixed alkyl chain lengths are particularly preferred.

The CPEs and RSEs for use in the invention include those recited in the following examples, including sucrose pentalaurate, sucrose tetraoleate, sucrose pentaerucate,

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sucrose tetraerucate, and sucrose pentaoleate. Suitable materials include some of the Ryoto series available from Mitsubishi Kagaku Foods Corporation.

- 5 Other nonionic fabric softening agents that may be used in the compositions include pentaerythritol esters, and sorbitan esters, mono, di and triglycerides, ester oils, mineral oils, fatty acids, fatty alcohols and alkyl polyglycosides.

10

The fatty acid may be a C<sub>8</sub>-C<sub>24</sub> alkyl or alkenyl monerocarboxylic acid. Preferably the fatty acid is saturated. The fatty alcohols may have the same chain length as above.

15

Mixtures of any of the aforementioned types of nonionic fabric softening agents may be used.

- 20 The fabric softening agent is present in the composition preferably in total amount of 0.5% - 80%, by weight based upon the total weight of the composition, more preferably 0.5% - 50%, more preferably 1 - 30%, more preferably as 2 - 25%, eg 3 - 20%.

25 Anionic Surfactant

The anionic surfactant may be any suitable anionic surfactant conventionally used in laundry compositions.

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The anionic surfactant may be chosen from soap and non-soap anionic surfactants and mixtures thereof.

Many suitable detergent active compounds are available and  
5 are fully described in the literature, for example, in  
"Surface-Active Agents and Detergents", Volumes I and II, by  
Schwartz, Perry and Berch.

Anionic surfactants are well-known to those skilled in the  
10 art. Examples include alkylbenzene sulphonates,  
particularly linear alkylbenzene sulphonates having an alkyl  
chain length of C<sub>8</sub>-C<sub>15</sub>; primary and secondary alkylsulphates,  
particularly C<sub>8</sub>-C<sub>15</sub> primary alkyl sulphates; alkyl ether  
sulphates; olefin sulphonates; alkyl xylene sulphonates;  
15 dialkyl sulphosuccinates; and fatty acid ester sulphonates.  
Sodium salts are generally preferred.

The compositions preferably comprise 0.1% - 10% by weight  
anionic surfactant, more preferably 0.2% - 5%, most  
20 preferably 0.5% - 3.5%.

The weight ratio of fabric softening agent to anionic  
surfactant is preferably in the range 15:1 to 1:10, more  
preferably 10:1 to 1:5, most preferably 10:1 to 1:1.

25

#### Nonionic Emulsifier

The compositions may optionally further comprise nonionic  
emulsifiers. Any nonionic emulsifier conventionally used in  
30 laundry compositions may be used e.g. nonionic ethoxylated

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surfactants have an HLB of from about 10 to about 20. It is advantageous if the surfactant alkyl group contains at least 12 carbon atoms. If present the nonionic surfactant may be used in amounts of 0.1 - 10% by weight, preferably 0.2 - 5%.

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Cationic polymers

- The compositions further comprise cationic polymers. Any suitable cationic polymers may be used according to the invention. The cationic polymer is believed to act as a 'bridging' polymer and aids deposition of the emulsified fabric softener particle onto the surface of the fabric being treated.
- 10 Suitable cationic polymers include cationic guar polymers and their derivatives (eg the Jaguar series of polymers available from Rhodia), cationic cellulose polymers and their derivatives (eg the Celquat series of polymers available from National Starch and Chemical Ltd and the
- 15 Ucare series of polymers available from Amerchol), cationic starches such as potato starch (eg the SoftGel and Solvitose series of polymers available from Avebe and the C\*bond polymers series from Cerestar), and cationic chitosan and derivatives. Mixtures of such polymers may also be used.
- 20 Any of the cationic polymers recited in the following examples are suitable for use in the compositions of the invention.
- 25 The compositions preferably comprise 0.01-5% by weight of the cationic polymer, more preferably 0.05-4.5%, most preferably 0.1-3.5%, eg 0.1-3%.
- 30 The compositions comprise no more than 1% by weight in total of non-polymeric cationic surfactant and/or cationic fabric



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softening compounds. Preferably the compositions are substantially free of said cationic materials.

If a cationic surfactant or cationic softening compound, eg  
5 a quaternary ammonium compound, is present in the composition it is preferred that it is present in an amount of 0.75% by weight or less, preferably 0.5% or less such as 0.2% by weight or less.

10 In the compositions the weight ratio of the softening agent to the cationic polymer is preferably within the range 100:1 to 1:1, preferably 40:1 to 1:1, e.g. 10:1 to 1:1.

In the compositions the weight ratio of the softening agent  
15 to the total amount of anionic surfactant and cationic polymer is preferably within the range 15:1 to 1:10, more preferably 10:1 to 1:5, most preferably 10:1 to 1:1.

#### Other Polymers

20

Nonionic polymers may optionally be present in the compositions in addition to the cationic polymers. Suitable nonionic polymers that may optionally be present include the Pluronic series of polymers available from BASF, dialkyl  
25 PEGs, cellulose derivatives as described in GB 213 730 (Unilever), hydroxyethyl cellulose, starch, and hydrophobically modified nonionic polyols such as Acusol 880/882 available from Rohm & Haas.

30 Anionic polymers may also be present in the composition.

Optional ingredients

- The compositions may also contain one or more optional ingredients, selected from oils, (such as vegetable oils and ester oils) electrolytes, non-aqueous solvents, pH buffering agents, perfumes, perfume carriers, fluorescers, colorants, hydrotropes, antifoaming agents, antiredeposition agents, polymeric and other thickeners, enzymes, optical brightening agents, opacifiers, anti-shrinking agents, anti-wrinkle agents, anti-spotting agents, germicides, fungicides, anti-oxidants, anti-corrosion agents, drape imparting agents, antistatic agents, sunscreens, colour care agents and ironing aids.
- 15 If the product is a liquid, a viscosity control agent may be included. Any viscosity control agent typically used with rinse conditioners is suitable, for example biological polymers such as Xanthum gum (Kelco ex Kelsan and Rhodopol ex Rhone-Poulenc). Synthetic polymers may also be used as
- 20 viscosity control agents eg polyacrylic acid, poly vinyl pyrrolidone, polyethylene, carbomers, polyethylene and polyethylene glycols.

- It is preferred that the compositions are substantially free of bleaches. It is especially preferred that the compositions are entirely free of bleaches.

**Product Form**

- 30 The compositions may be in any form conventionally used for fabric softening compositions for example, powder, paste,

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gel or liquid. It is preferred if the product is a liquid and especially an emulsion.

The compositions may be prepared by any suitable method.

One method (method A) is to dissolve the cationic polymer in water, optionally with heating to assist dissolution, and then add the anionic surfactant. The solution initially becomes cloudy but clears when the polymer / surfactant complex re-dissolves. After this point the nonionic softener is added.

10

Another method (method B) is to emulsify the nonionic fabric softener with the anionic surfactant and then to post-dose an aqueous solution of the cationic polymer to this emulsion. A further method (method C) is to solubilise the polymer in solution and then add the anionic surfactant/nonionic softener as a co-melt.

15

### Examples

The invention is illustrated by the following non-limiting examples. Further examples within the scope of the present invention will be obvious to the man skilled in the art.

Samples of the invention are denoted by a number and comparative samples are denoted by a letter.

25

### Example 1

Compositions 1 to 24 in the table below were prepared by method A. All % are by weight as the active ingredient.

30

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Compositions A and B were prepared by dissolving the anionic surfactant in water, followed by adding the nonionic softener and mixing the composition for 10 minutes.

Sample No.	Cationic Polymer Type and % by weight	Anionic surfactant type and % by weight	Fabric softener type and % by weight
1	A-0.05%	ABS-2%	ER290-4.5%
2	A-0.1%	ABS-2%	ER290-4.5%
3	B-0.1%	ABS-1%	ER290-4.5%
4	C-0.3%	ABS-1%	ER290-4.5%
5	C-0.4%	ABS-1%	ER290-4.5%
6	C-0.5%	ABS-1%	ER290-4.5%
7	C-0.7%	ABS-1%	ER290-4.5%
8	C-0.8%	ABS-1%	ER290-4.5%
9	C-0.9%	ABS-1%	ER290-4.5%
10	C-1.0%	ABS-1%	ER290-4.5%
11	C-1.5%	ABS-0.8%	ER290-4.5%
12	D-0.1%	ABS-1%	ER290-4.5%
13	D-0.2%	ABS-1%	ER290-4.5%
14	D-0.3%	ABS-1%	ER290-4.5%
15	E-0.1%	ABS-1%	ER290-4.5%
16	E-0.2%	ABS-1%	ER290-4.5%
17	E-0.3%	ABS-1%	ER290-4.5%
18	F-1.5%	SDS-0.75%	ER290-4.5%
19	F-2.0%	SDS-0.75%	ER290-4.5%
20	A-0.3%	ABS-3%	ER290-15%
21	C-3.0%	ABS-3%	ER290-15%
22	C-2.0%	G-0.50%	ER290-4.5%
23	C-1.5%	G-2.0%	ER290-2.5%
24	C-1.5%	G-3.5%	ER290-1.0%
A	--	G-2.0%	ER290-2.3%
B	--	G-3.0%	ER290-1.5%

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Polymer type A was Jaguar C13-S from Rhodia; a cationic guar gum.

Polymer type B was Jaguar C162 from Rhodia; a cationic guar gum.

5 Polymer type C was Softgel BDA from Avebe; a cationic potato starch.

Polymer type D was Ucare JR125 from Amerchol; a cationic cellulose.

10 Polymer type E was Ucare JR400 from Amerchol; a cationic cellulose.

Polymer type F was Solvitose from Avebe; a cationic starch.

ABS is sodium dodecyl benzene sulphonate from Aldrich.

SDS is sodium dodecyl sulphate from Aldrich.

15

G is sodium cocoyl isothionate from Akzo

ER290 is Ryoto ER290 (sucrose tetraerucate) available from Mitsubishi Kagaku Foods Corporation.

20

The compositions were all homogeneous in appearance. The particles formed from the cationic polymer, anionic surfactant and fabric softener had a net overall negative charge.

25

#### Example 2

Samples 6-10, 22, A and B and a commercial dilute fabric softening composition, C, were tested for fabric softening  
30 ability. To simulate the effects of carryover of anionic surfactant from the wash various amounts of 1% by weight

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solution of alkyl benzene sulphonate were added to the rinse liquor to determine how resistant the compositions are to such anionic carryover.

#### 5 Softness Evaluation

Softening performance was evaluated by adding 0.1g of fabric softening compound (2ml of a 5% a.d. dispersion for liquids) to 1 litre of tap water, at ambient temperature in a  
10 tergotometer. Three pieces of terry towelling (8cm x 8cm, 40g total weight) were added to the tergotometer pot. The cloths were treated for 5 minutes at 65 rpm, spin dried to remove excess liquor and line dried overnight and conditioned at 21°C/65°C.

15

Softening of the fabric was assessed by an expert panel of 4 people using a round robin paired test protocol. Each panel member assessed four sets of test cloths. Each set contained one cloth of each sample under evaluation. Panel  
20 members were asked to assess softness on a nine point scale. In the table below a score of 1 represents a very soft fabric and 9 represents a very harsh fabric. Softness scores were calculated using an 'Analysis of Variance technique.

25

The softening results are given below.

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Sample No.	No of mls of 1% ABS added to replicate increasing levels of anionic carryover				
	0	1	2	3	4
6	4.75	4.25	3.75	3	4.25
7	-	3.0	2.75	2.5	2
8	-	2.5	2.75	2.75	3
9	-	3.0	3.5	2.75	3.5
10	3.75	4	3.75	3.75	4.0
22	3.9	-	-	4.6	-
A			5.9		
B			6.8		
C <sup>a</sup>			3.9		

<sup>a</sup> dilute Comfort (commercially available June 1999)

5 The above results demonstrate that the compositions of the invention provide excellent softening results at various levels of simulated anionic carryover. The compositions also did not significantly decrease the absorbency of the treated fabric.

10

The results also show that, at 2mls carryover, the compositions of the invention provided significantly better softening than the comparative composition which lacked a cationic polymer, and better softening than a commercially available dilute fabric softening composition

15

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Example 3

A fully formulated fabric softening composition as according to the present invention was prepared as below:

5

	% by weight
ABS	1.0
Sucrose ester oil	4.5
Softgel BDA	1.0
Dye	0.0025
Perfume	0.3
Minors	0.02
Water	balance

The sucrose ester oil was Ryoto ER290 available from Mitsubishi Kagaku Foods Corporation. ABS and Softgel BDA

10 are described above.



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Example 4

A second fully formulated fabric softening composition was prepared as below:

5

	% by weight
ABS	0.4
Sucrose ester oil	15.4
Softgel BDA	2.0
Dye	0.0025
Perfume	0.96
Nonionic emulsifier	1.6
Minors	0.15
Water	Balance

The sucrose ester oil was Ryoto ER290, described above. The nonionic emulsifier was coco alcohol (15 EO).

10

The composition was prepared as follows:

An aqueous solution comprising 20wt% of the nonionic emulsifier and the ABS (in a 4:1 weight ratio) was firstly  
15. mixed with 30.84g of the sucrose ester oil and perfume under stirring, to form a water in oil emulsion.

Then 50g of the Softgel BDA at 8wt% was stirred into the emulsion followed by 98.36g of cold demineralised water  
20 (also added under stirring). Finally the dye (0.5g) and the minors were added and stirring was continued for a further 10 minutes.

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The composition was monitored for 12 weeks, during which time it remained stable and homogenous.

### Example 5

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The table below shows the T2 NMR solid:liquid ratio of CPE's and RSE's used according to the present invention. The ratios were measured at 20°C. The degree of esterification /etherification is stated.

10

Material	Solid:liquid ratio at 20°C	Degree and % of esterification		Physical form
Ryoto 0-170 <sup>1</sup>	0:100	5/8	62.5%	Liquid
Ryoto ER-290 <sup>2</sup>	0:100	4/8	50%	Liquid
Ryoto ER-190 <sup>3</sup>	0:100	5/8	50%	Soft solid
Ryoto POS-135 <sup>4</sup>	30:70	5/8	62.5%	Soft solid
Ryoto L-195 <sup>5</sup>	43:57	5/8	62.5%	Liquid
Sucrose tetraoleate	0:100	4/8	50%	Liquid
Sucrose octaoleate	0:100	8/8	100%	Liquid

1 = sucrose pentaoleate

2 = sucrose tetraerucate

3 = sucrose pentaerucate

15 4 = sucrose pentatallowate

5 = sucrose pentalaurate

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Example 6; Fabric Softening Compositions

The following example shows compositions according to the invention comprising various nonionic fabric softening agents.

The compositions were prepared by firstly adding the cationic polymer (hot) to water followed by adding the molten nonionic softener/anionic surfactant mixture thereto. The only exception to this was sample 1, where the subsequent order of addition was SLES followed by the sucrose monoester (coco/tallow chains).

In the samples where either Na soap or K soap is present (samples 2-5), this was achieved through in-situ neutralisation of HT fatty acid by either NaOH or KOH respectively. In these cases the NaOH or the KOH was added after the polymer and before the co-melted actives.

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Ingredient	sample 1	sample 2	sample 3	sample 4	sample 5
Hardened tallow fatty acid <sup>2</sup>	0	2.17%	1.58%	1.59%	1.56%
Hardened tallow Na soap	0	2.51%	1.97%	1.95%	0
Hardened tallow fatty alcohol <sup>3</sup>	0	0	1.1%	0	1.1%
Hardened tallow K soap	0	0	0	0	2.09%
Sucrose monoester (coco/tallow chains) <sup>1</sup>	4.5%	0	0	1.1%	0
Na laureth ether sulphate (SLES) <sup>4</sup>	0.5%	0.5%	0.5%	0.5%	0.5%
Cationic potato starch <sup>5</sup>	1%	1%	1%	1%	1%
Water	to 100%	to 100%	to 100%	to 100%	To 100%
Softness score	4.75	4.50	4.25	4.00	4.00

1 = Sucrose monoester (coco/tallow chains) is available as Tegosoft PSE1419 (ex Goldschmidt AG)

5 2 = Hardened tallow fatty acid is available as Pristerine 4916 (ex Unichema)

- 25 -

- 3 = Hardened tallow fatty alcohol is available as Laurex CS (ex Albright & Wilson)
- 4 = Na laureth ether sulphate (SLES) is available as Elfan NS2436 (ex Akzo-Nobel)
- 5 5 = Cationic potato starch is available as Softgel BDA (ex Avebe)

Example 7; Measurement of Zeta potential

- 10 The Zeta potential of the following example was measured on a Malvern Instrument Zeta Sizer.

0.75 wt% Softgel BDA (ex Avebe)

0.75% wt% SDS (ex Adrich)

- 15 4.5% wt% ER290 (ex Mitsubishi Kagaku)

Softgel BDA, SDS and ER290 are as described above.

- The average zeta potential was minus 25.2 demonstrating that fabric softening particles formed from a nonionic fabric
- 20 softener, an anionic surfactant and a cationic polymer have an overall net negative charge.

CLAIMS

1. A fabric softening composition comprising:

5 (i) at least one nonionic fabric softening agent, and

(ii) at least one anionic surfactant, and

(iii) at least one cationic polymer

10

wherein the particles formed from i), ii) and iii) have a net negative charge and the composition comprises no more than 1% by weight non-polymeric cationic surfactant and/or cationic fabric softening compounds.

15

2) A composition according to claim 1 wherein the nonionic fabric softener (i) comprises a liquid or soft solid derivative of a cyclic polyol (CPE) or of a reduced saccharide (RSE) resulting from 35 to 100% of the hydroxyl groups in the polyol or saccharide being esterified or etherified, the CPE or RSE having 2 or more ester or ether groups independently attached to a C<sub>8</sub>-C<sub>22</sub> alkyl or alkenyl chain.

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25 3) A composition according to claim 2 wherein the CPE or RSE contains at least 35% tri or higher esters.

4) A composition according to any one of the preceding claims wherein the cyclic polyol or reduced saccharide has 40-80%, preferably 45 - 75%, of the hydroxyl groups esterified and/or etherified.

30

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5) A composition according to any one of the preceeding claims wherein the CPE or RSE has 3 or more ester or ether groups, preferably 4 or more.

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6) A composition according to any one of the preceeding claims wherein the CPE or RSE is derived from a monosaccharide or disaccharide.

10 7) A composition according to any one of the preceding claims comprising 0.5% - 50% by weight of the fabric softening agent, preferably 1% - 30%.

15 8) A composition according to any one of the preceding claims comprising 0.1% - 10% by weight of the anionic surfactant, preferably 0.2% - 5%.

9) A composition according to any one of the preceding claims wherein the cationic polymer is selected from  
20 cationic guar derivatives, cationic cellulose derivatives, cationic starch derivatives and, cationic chitosan derivatives or mixtures thereof.

10) A composition according to any one of the preceding  
25 claims comprising 0.01% - 5% by weight of the cationic polymer, preferably 0.1-3%.

11) A composition according to any one of the preceding claims wherein the weight ratio of softening agent to the  
30 total amount of anionic surfactant and cationic polymer is within the range 15:1 to 1:10, preferably 10:1 to 1:1.

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12) A composition according to any preceding claim which is a liquid, preferably an emulsion.

13) A method of depositing a nonionic fabric softening agent onto fabric from a fabric softening composition, comprising emulsifying the softening agent with an anionic surfactant and a cationic polymer to form a particle having an overall net negative charge in the composition and treating said fabric with said composition.

10

14) A method of depositing a nonionic fabric softening agent onto fabric from a fabric softening composition, comprising emulsifying the softening agent with an anionic surfactant and then post-dosing an aqueous solution of a cationic polymer to form a particle having an overall negative charge on the composition and treating said fabric with said composition.

15) A method according to claim 13 or claim 14 wherein the fabric softening agent is as defined in any one of the preceding claims.

20

16) A method according to either claim 14 or claim 15 wherein the cationic polymer is as defined in claim 9.

25



# INTERNATIONAL SEARCH REPORT

International Application No

PCT/GB 00/01706

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C11D1/83 C11D3/22 //C11D1:12,C11D1:14,C11D1:22

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C11D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 0 093 601 A (UNILEVER NV) 9 November 1983 (1983-11-09)	1,9, 11-13, 15,16 1-16
Y	page 4, line 15 -page 7, paragraph 4 page 8, line 33 -page 10, line 18 examples 12,13 claims 1,11	
Y	WO 98 16538 A (UNILEVER) 23 April 1998 (1998-04-23) cited in the application page 5, line 25 -page 9, line 26 page 14, line 28 -page 15, line 30 examples 50-53 claims 1-8	1-16

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

### \* Special categories of cited documents:

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- "P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

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"&" document member of the same patent family

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## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

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